## Supplementary Online Material Photoinduced Proton Coupled Electron Transfer in 2-(2'-Hydroxyphenyl)-Benzothiazole

Sandra Luber<sup>1†</sup>, Katrin Adamczyk<sup>2</sup>, Erik T. J. Nibbering<sup>2\*</sup>, and Victor S. Batista<sup>1\*</sup>

<sup>1</sup>Department of Chemistry, Yale University, New Haven, Connecticut, 06520-8107, USA; <sup>2</sup>Max-Born-Institut für Nichtlineare Optik und Kurzzeitspektroskopie, Max-Born-Strasse 24, 12489 Berlin-Adlershof, Germany

\* nibberin@mbi-berlin.de; victor.batista@yale.edu

<sup>†</sup> present address: University of Zurich, Winterthurerstrasse 190, 8057 Zurich, Switzerland, e-mail: s.luber@pci.uzh.ch.

## **Detailed Computational Methodology**

For all calculations which include solvent effects, the quantum chemical package GAUSSIAN [1] has been employed whereas calculations without consideration of any solvent effect have been performed with the quantum chemical package TURBOMOLE1 [2] [except of Fig. 2 (a)]. For the ground-state structure optimization (if not noted otherwise) and IR spectrum calculation, density functional theory (DFT) was used with the BP86 density functional [3,4] and Ahlrichs TZVP [5] basis set. For excited-state structure optimizations and IR spectra calculations, time-dependent DFT (TD-DFT) with the B3LYP density functional [6,7], as implemented in GAUSSIAN 09 and TURBOMOLE (Version 6.2), respectively, and the TZVP or TZVPP [5,8] basis sets were used as well as the Configuration Interaction Singles (CIS) method with the TZVP basis set. The B3LYP density functional in GAUSSIAN 09 employs the Vosko, Wild, and Nusair (VWN) correlation functional(III) [9] whereas the B3LYP density functional implementation in Turbomole version 6.2 uses the VWN(V) correlation functional [9]. In addition, IR excitedstate spectra were obtained with the M06-2X [10] and CAM-B3LYP [11] density functionals. The electronic transition dipole moments of cis-enol from the ground to the first electronically excited state were obtained with TDDFT employing B3LYP and the TZVP or TZVPP basis set. The solvent tetrachloroethene is considered via the polarizable continuum model (PCM) using the integral equation formalism variant as implemented in GAUSSIAN 09. The normal modes, frequencies, and IR intensities as well as the anisotropy angles of certain normal modes were obtained with a modified and extended version of SNF[12,13] in the harmonic approximation. GAUSSIAN 09 and TURBOMOLE, respectively, were employed for the calculation of electronic energy gradients and electric dipole moments for distorted structures of the molecule of interest. This data was collected by SNF that evaluates the normal modes and IR intensities. For the differentiation, a three-point central difference formula [14] was applied and the step length for the differentiation was chosen to be 0.01 bohr. The electrostatic-potential-fitted (ESP) charges were obtained employing the Merz-Singh-Kollman [15,16] scheme as implemented in GAUSSIAN 09. Molecular structures and normal modes were visualized using the programs VMD [17] and JMOL [18], respectively.

Figure 1: IR spectra of HBT in the S<sub>1</sub>-state with cis-keto<sup>\*</sup> configuration, calculated with (a) TDDFT (B3LYP/TZVP), (b) TDDFT(B3LYP/TZVP/PCM)/CIS(TZVP/PCM), (c) TDDFT (CAM-B3LYP/TZVP/PCM), and (d) TDDFT (M06-2X/TZVP/PCM) are compared with the transient IR spectrum measured at 100 ps pulse delay (d).



## **Comparison of Experimentally Measured and Calculated Anisotropies**

Table 1: Measured and calculated [TDDFT(B3LYP/TZVP/PCM)] anisotropy angles  $\Theta$  of selected experimental bands / calculated normal modes  $\Omega$  of cis-keto (in brackets: trans-keto\*) in the first electronically excited state.

$\Omega \ ({\rm cm}^{-1})$		Θ (°)	
Exp.	Theory	Exp.	TDDFT
1305	1308 (1305)	$29 \pm 5$	39(4)
1397	626 + 794 (625 + 770)	90 - 15	90 (90)
1439	1422(1421)	$77 \pm 13$	52(31)
1475	1475(1475)	90 - 15	68 (86)
1475	1482 (1491)	90 - 15	62(31)
1535	1535 (1520)	$23 \pm 2.5$	16 (29)

Table 2: Measured and calculated anisotropy angles  $\Theta$  of selected experimental bands / calculated normal modes  $\Omega$  of cis-keto in the first electronically excited state; the calculated values were obtained with normal modes employing TDDFT(B3LYP/TZVP/PCM) and electric-dipole moments computed with CIS(TZVP/PCM).

$\Omega \ ({\rm cm}^{-1})$		Θ (°)		
Exp.	Theory	Exp.	(TDDFT/CIS)	
1305	1308	$29 \pm 5$	12	
1397	(703+735)	90 - 15	90	
1439	1142	$77 \pm 13$	36	
1475	1475	90 - 15	68	
1475	1482	90-15	38	
1535	1535	$23 \pm 2.5$	12	

## References

- [1] Frisch, M. J. et al. "GAUSSIAN 09, Revision A.1", Gaussian Inc. Wallingford CT 2009.
- [2] Ahlrichs, R.; Bär, M.; Häser, M.; Horn, H.; Kölmel, C. Electronic structure calculations on workstation computers: The program system Turbomole *Chem. Phys. Lett.* **1989**, *162*, 165-169.
- [3] Becke, A. D. Density-functional exchange-energy approximation with correct asymptotic behavior *Phys. Rev. A* 1988, 38, 3098-3100.
- [4] Perdew, J. P. Density-functional approximation for the correlation energy of the inhomogeneous electron gas *Phys. Rev. B* 1986, 33, 8822-8824.
- [5] Schäfer, A.; Huber, C.; Ahlrichs, R. Fully optimized contracted Gaussian basis sets of triple zeta valence quality for atoms Li to Kr J. Chem. Phys. 1994, 100, 5829-5835.

- [6] Becke, A. D. Density-functional thermochemistry. III. The role of exact exchange J. Chem. Phys. 1993, 98, 5648-5652.
- [7] Lee, C.; Yang, W.; Parr, R. G. Development of the Colle–Salvetti correlation-energy formula into a functional of the electron density *Phys. Rev. B* 1988, 37, 785-789.
- [8] Dunning, Jr., T. H. Gaussian basis sets for use in correlated molecular calculations. I. The atoms boron through neon and hydrogen J. Chem. Phys. 1989, 90, 1007-1023.
- [9] Vosko, S. H.; Wilk, L.; Nusair, M. Accurate spin-dependent electron liquid correlation energies for local spin density calculations: a critical analysis *Can. J. Phys.* 1980, 58, 1200-1211.
- [10] Zhao, Y.; Truhlar, D. G. The M06 suite of density functionals for main group thermochemistry, thermochemical kinetics, noncovalent interactions, excited states, and transition elements: two new functionals and systematic testing of four M06-class functionals and 12 other functionals *Theor. Chem. Acc.* 2008, 120, 215-241.
- [11] Yana, T.; Tew, D.; Handy, N. A new hybrid exchange–correlation functional using the Coulomb-attenuating method (CAM-B3LYP) *Chem. Phys. Lett.* **2004**, *393*, 51-57.
- [12] Neugebauer, J.; Herrmann, C.; Luber, S.; Reiher, M. SNF 4.0 a program for the quantum chemical calculation of vibrational spectra, URL: http://www.reiher.ethz.ch/software/snf.
- [13] Neugebauer, J.; Reiher, M.; Kind, C.; Hess, B. A. Quantum Chemical Calculation of Vibrational Spectra of Large Molecules — Raman and IR Spectra for Buckminsterfullerene J. Comput. Chem. 2002, 23, 895-910.
- [14] Bickley, W. G. Formulae for numerical differentiation Math. Gaz. 1941, 25, 19-27.
- [15] Singh, U. C.; Kollman, P. A. An approach to computing electrostatic charges for molecules J. Comput. Chem. 1984, 5, 129-145.
- [16] Besler, B. H.; Merz, K. M. J.; Kollman, P. A. Atomic charges derived from semiempirical methods J. Comput. Chem. 1990, 11, 431-439.

- [17] Humphrey, W.; Dalke, A.; Schulten, K. VMD- Visual Molecular Dynamics J. Molec. Graphics 1996, 14, 33-38.
- $[18]~2003~{\rm Jmol}$  a free, open source molecule viewer, URL: http://jmol.sourceforge.net.