

Supporting Online Material for

Theoretical EXAFS studies of a model of the oxygen-evolving complex of photosystem II obtained with the quantum cluster approach.

Xichen Li^{1,2,}, Eduardo M Sproviero⁴, Ulf Ryde³, Victor S Batista⁴, Guangju Chen¹.*

1. College of Chemistry, Beijing Normal University, 100875, Beijing, China. 2. Department of Physics, ALBA NOVA, and Department of Biochemistry and Biophysics, Arrhenius Laboratory, Stockholm University, SE-106 91 Stockholm, Sweden. 3. Department of Theoretical Chemistry, Lund University, Chemical Center, P.O. Box 124, S-221 00 Lund, Sweden. 4. Department of Chemistry, Yale University, PO Box 208107, New Haven, CT 06520-8107, USA.

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Corresponding author:

*To whom correspondence should be addressed. E-mail: xli@fysik.su.se.

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Cartesian coordinates of the refined cluster model

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- Calculated polarized EXAFS spectra in k -space.
- Comparison of the experimental and the calculated polarized EXAFS spectra with certain atoms removed.
- Comparison of calculated isotropic EXAFS spectrum.
- Individual atomic contributions to whole isotropic spectrum.
- Scatter contributions to the isotropic EXAFS spectrum of Mn1 based on path analysis.
- Scatter contributions to the isotropic EXAFS spectrum of Mn2 based on path analysis.
- Scatter contributions to the isotropic EXAFS spectrum of Mn3 based on path analysis.
- Scatter contributions to the isotropic EXAFS spectrum of Mn4 based on path analysis.

References

Cartesian coordinates of the refined cluster model

Mn	29.6201456	38.2278094	66.4149376
Mn	26.5717874	40.1116494	69.3324181
Mn	28.8366467	38.0116989	68.9520148
Mn	28.5872679	39.7084067	71.0818856
O	28.5516491	39.2752043	67.7191088
O	29.8323303	39.1703234	69.9200289
O	28.0225735	41.0745615	69.9802001
O	27.4367903	38.6398682	70.1038117
Ca	29.9276503	41.2324733	68.4401020
C	33.3151681	40.0124718	64.0344725
H	34.0154229	39.2058538	64.1938794
H	33.5463982	40.6512998	63.1949290
H	32.3692763	39.5250805	63.7724428
C	33.0943499	40.8240235	65.3252551
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H	34.0687456	41.1313814	65.8013652
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O	31.9666979	40.7423817	67.4565768
O	31.7718300	38.9586604	66.0889967
C	24.3931826	46.2892876	68.4847244
H	23.6554789	46.0724808	67.7263122
H	24.7089561	47.3202042	68.5471610
H	23.9123224	46.0597218	69.4465201
C	25.5612800	45.2974688	68.2882020
H	26.3188760	45.4206408	69.0938854
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C	23.9450124	39.1673755	67.7681376
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H	26.2789201	33.7010611	68.2404710
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C	26.0923255	41.0622870	74.8555534
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C	25.2553518	40.9676037	73.5733911
H	24.8148671	41.9616622	73.3313779
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H	34.4072957	40.8574593	70.0114833
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H	25.6914074	33.0766281	73.6315154
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O	29.1301964	39.3923223	65.0295008
H	29.6657364	39.1749026	64.2314144
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H	30.5802450	35.8318505	65.3652692
O	27.8964579	35.2765989	63.6288034
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O	26.0939728	36.4898483	61.6120870
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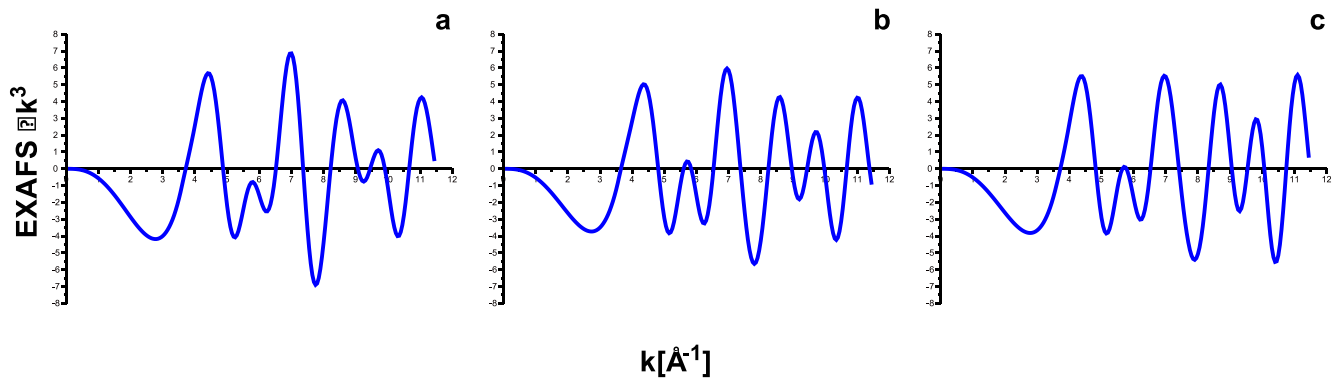


Figure S1. Calculated polarized EXAFS spectra in k -space for the refined cluster model structure. Unfortunately, we cannot compare to the experimental spectrum in k space, because it has not been published.

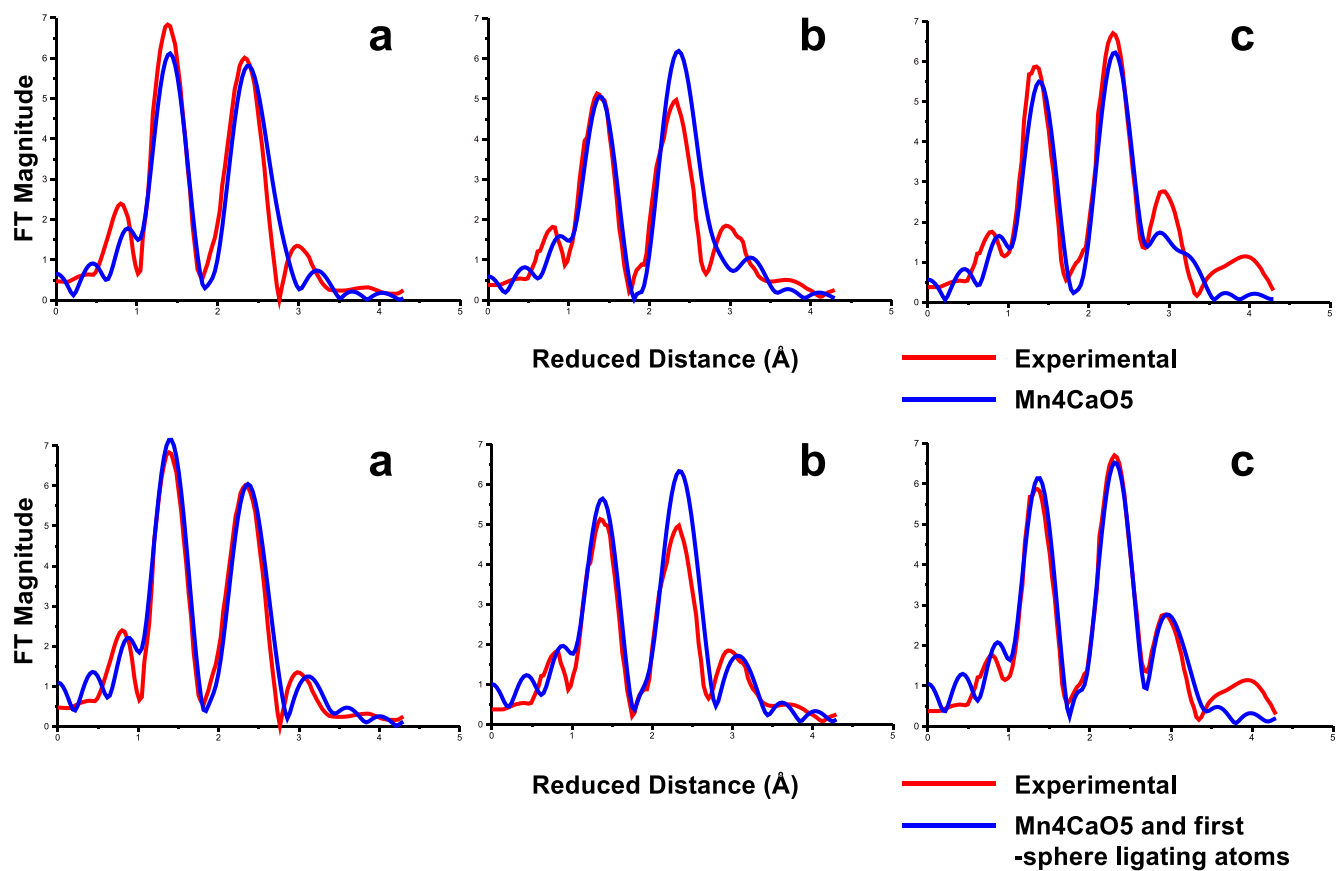


Figure S2. Comparison of the experimental (red) and the calculated polarized EXAFS spectra of the refined cluster model (blue) with certain atoms removed. Upper panel: only Mn_4CaO_5 was retained. Bottom panel: Mn_4CaO_5 and the first-sphere ligating atoms were retained. It can be noted that the general features of the spectra are reproduced already with Mn_4CaO_5 , but the finer details require inclusion also of the first-sphere ligating atoms.

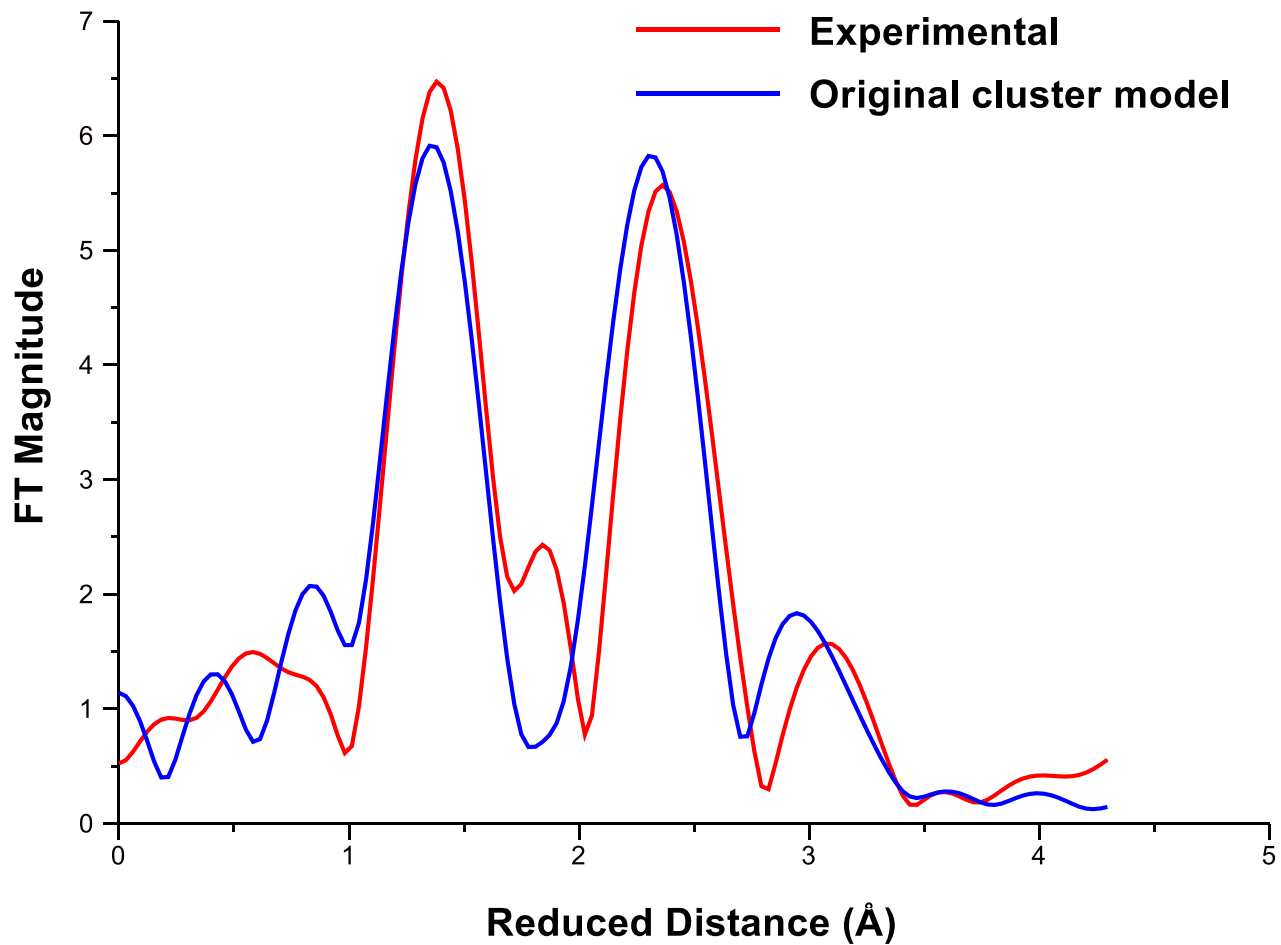


Figure S3. Comparison of calculated isotropic EXAFS spectrum for the refined cluster model (blue) and one set of experimental data¹ (red) (first converted from k -space into R -space).

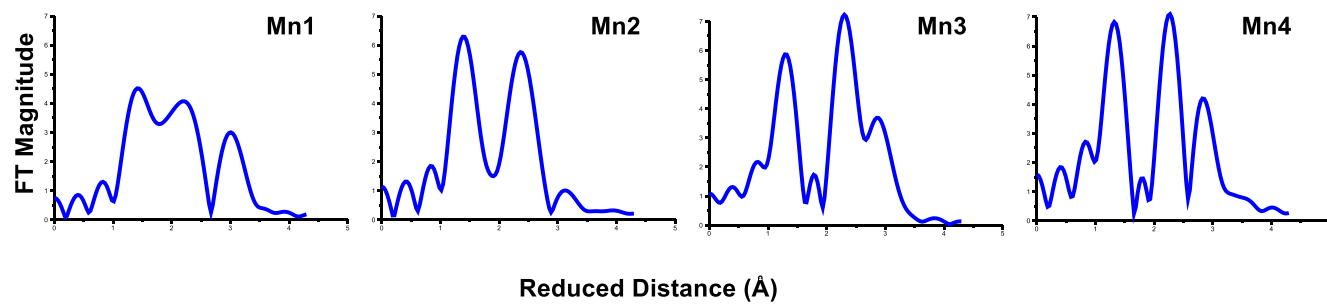


Figure S4. Individual contributions from the four Mn ions to the calculated isotropic EXAFS spectrum.

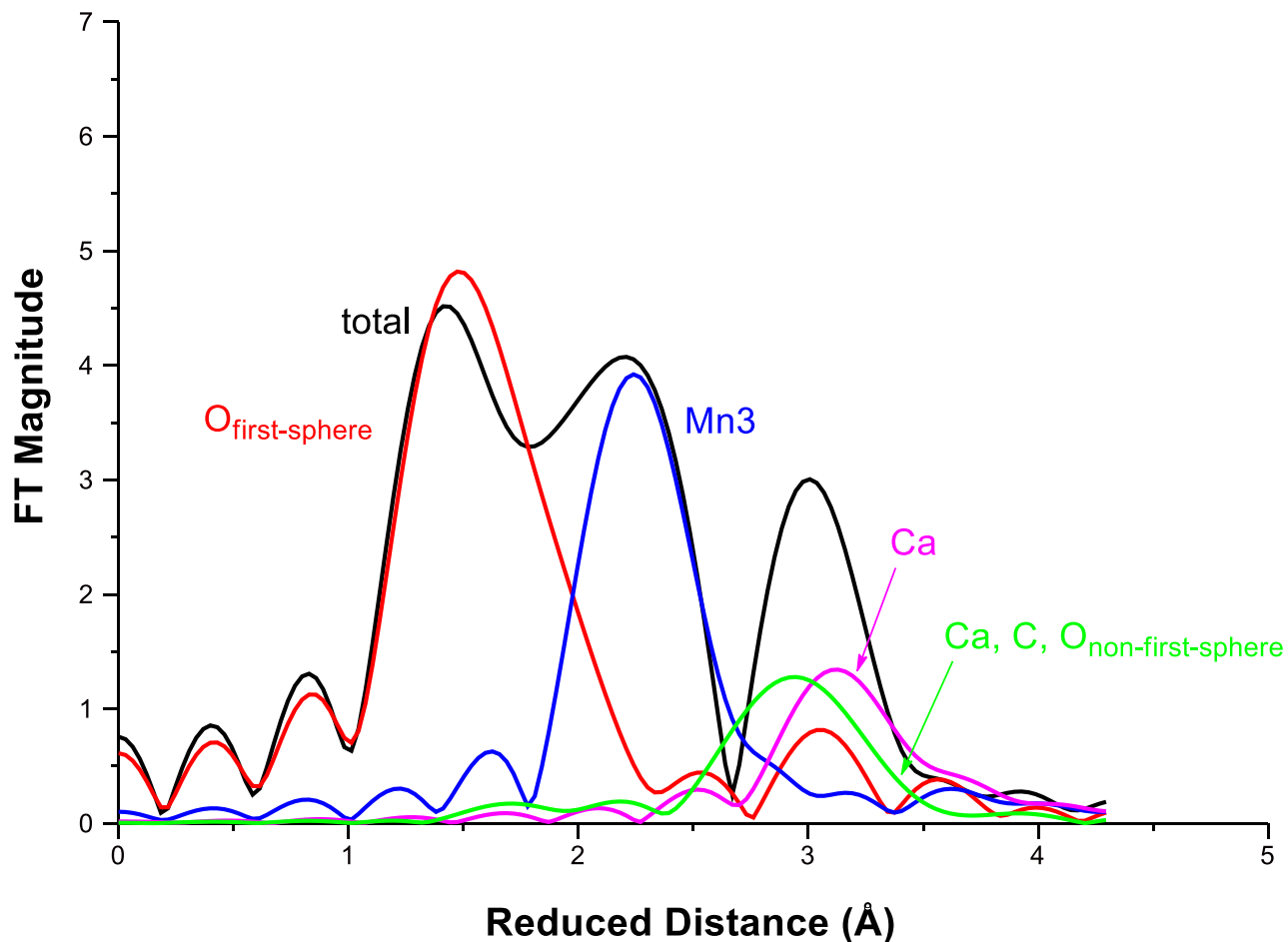


Figure S5. Path analysis of the Mn1 contribution to the isotropic EXAFS spectrum of the cluster model. The black curve is the total contribution, shown in Figure S4a, whereas the colored curves show the contributions from paths involving certain atoms. The first peak (around $R = 1.4 \text{ \AA}$) is caused by paths involving Mn1 and its first-sphere ligating oxygen atoms (red). The second peak (around $R = 2.3 \text{ \AA}$) is mainly due to paths involving Mn1 and the other Mn ions (blue). The third peak (around $R = 3.1 \text{ \AA}$) is mainly due to paths involving Mn1 and Ca, whereas the contribution from C and other O atoms is minor.

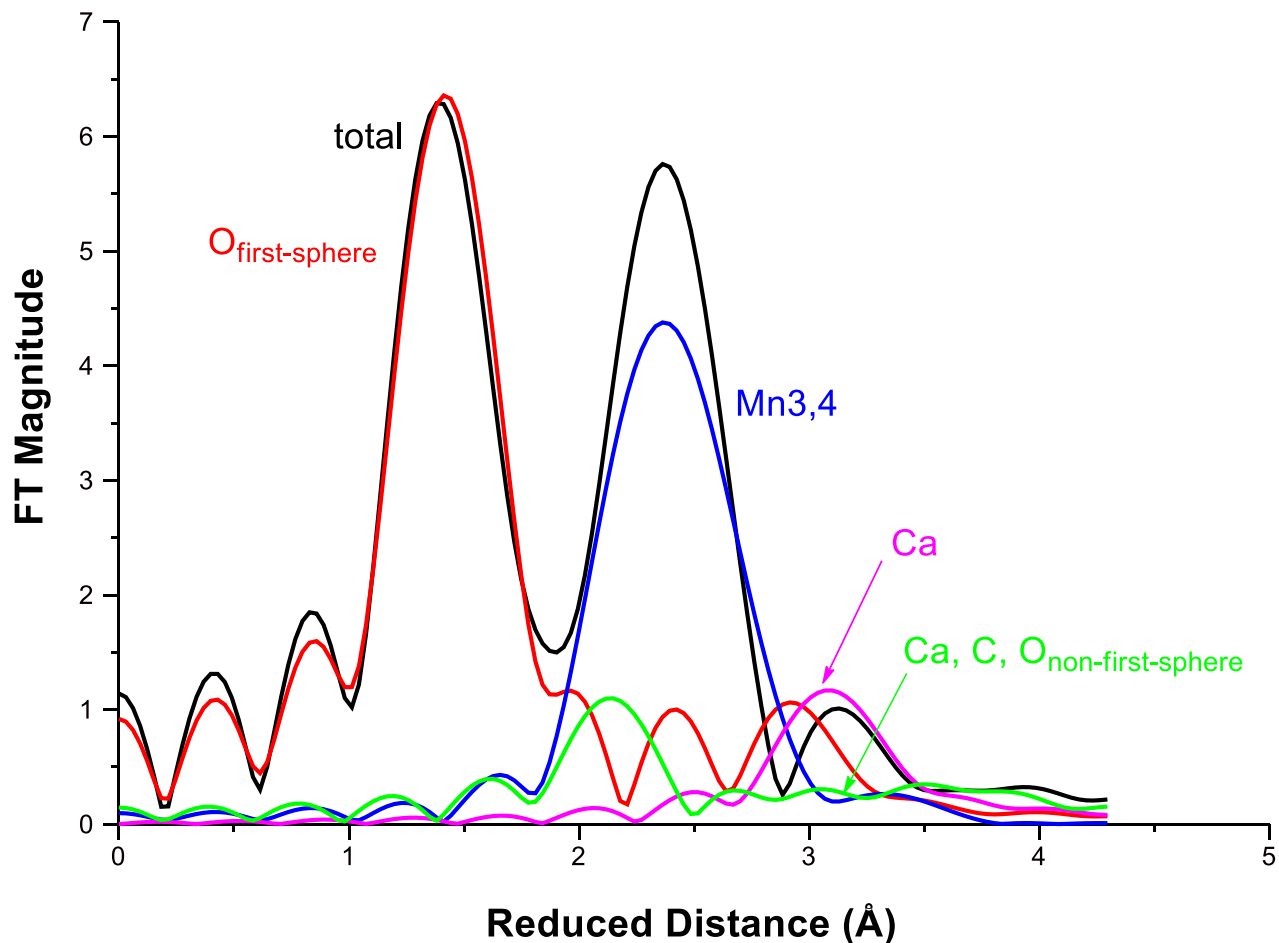


Figure S6. Path analysis of the Mn₂ contribution to the isotropic EXAFS spectrum of the cluster model. The first peak is caused by paths involving Mn₂ and its first-sphere ligating oxygen and nitrogen atoms (red). The second peak is mainly due to paths involving Mn₂ and the Mn₃ or Mn₄ ions (blue). The third peak is mainly due to paths involving Mn₂ and Ca, C, and the other oxygen atoms.

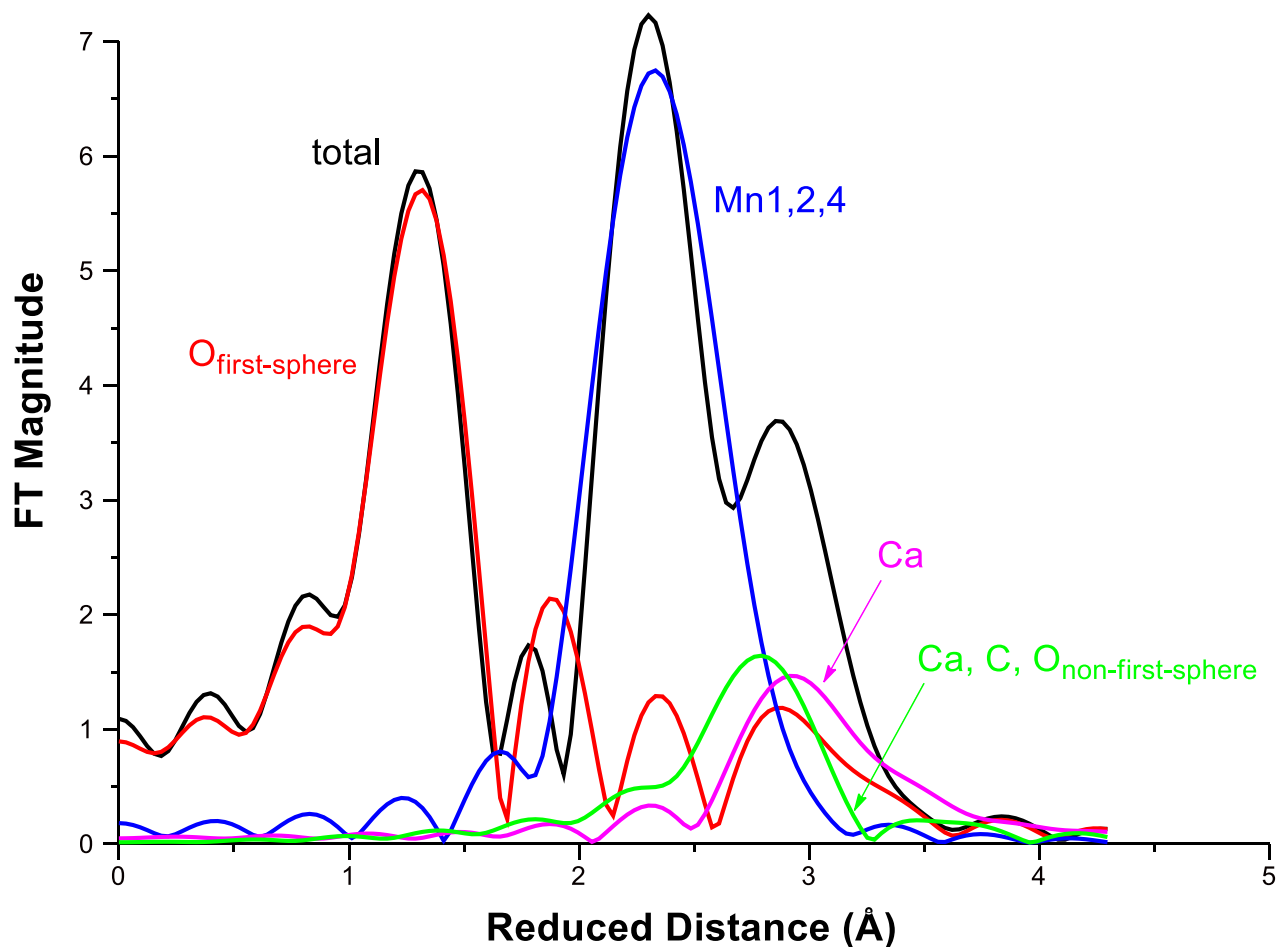


Figure S7. Path analysis of the Mn³⁺ contribution to the isotropic EXAFS spectrum of the cluster model. The first peaks ($R < 1.5$ Å) are caused mainly by paths involving Mn³⁺ and its first-sphere ligating oxygen atoms (red). The second peak is mainly due to paths involving Mn³⁺ and the other Mn ions (blue). The third peak is mainly due to paths involving Mn³⁺ and Ca, C, and the other oxygen atoms.

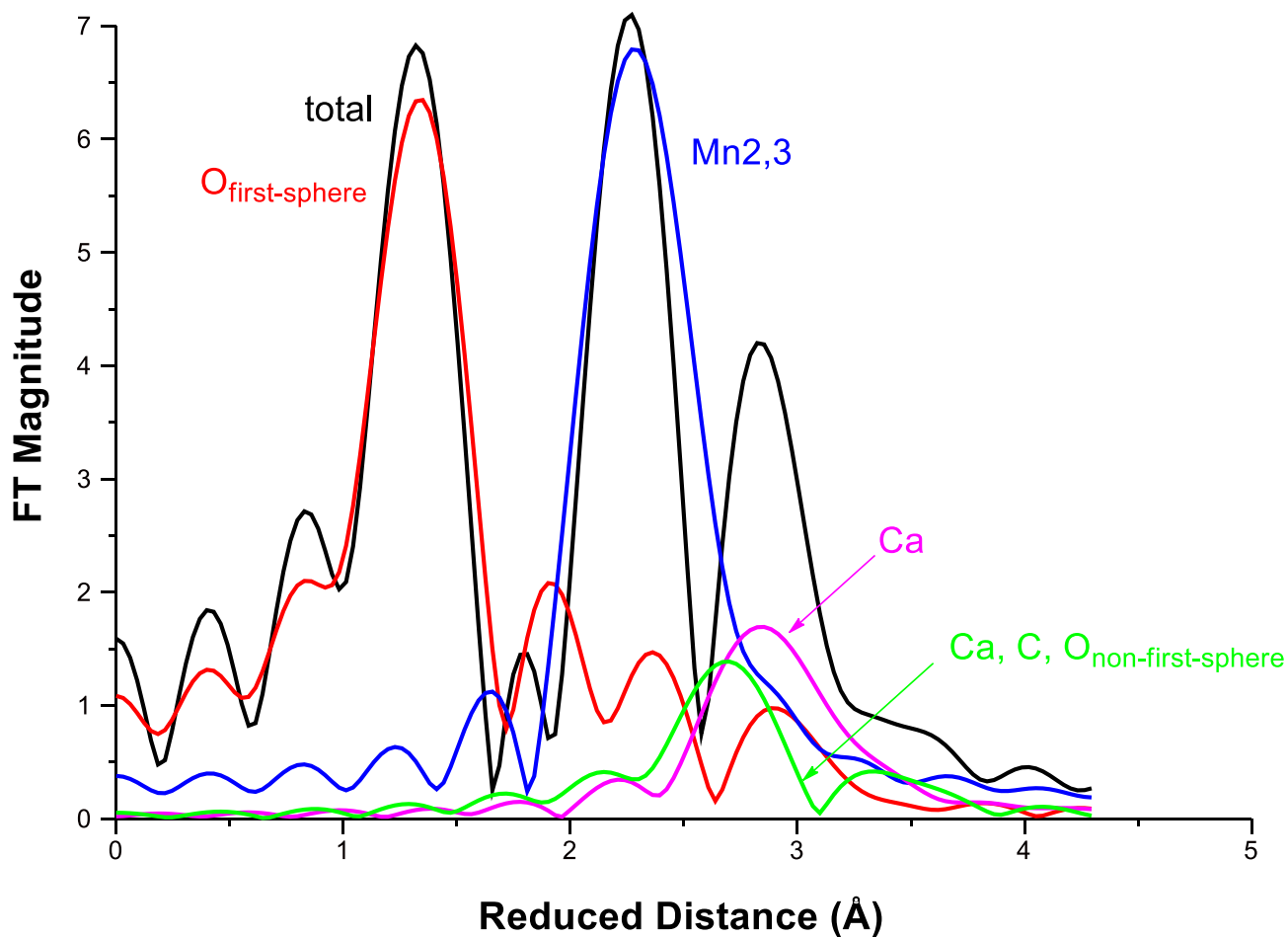


Figure S8. Path analysis of the Mn4 contribution to the isotropic EXAFS spectrum of the cluster model. The first peak is caused by paths involving Mn4 and its first-sphere ligating oxygen atoms (red). The second peak is mainly due to paths involving Mn4 and the Mn2 or Mn3 ions (blue). The third peak is mainly due to the paths involving Mn4 and Ca,C, and the other oxygen atoms.

References

1. Haumann, M.; Müller, C.; Liebisch, P.; Iuzzolino, L.; Dittmer, J.; Grabolle, M.; Neisius, T.; Meyer-Klaucke, W.; Dau, H., Structural and Oxidation State Changes of the Photosystem II Manganese Complex in Four Transitions of the Water Oxidation Cycle ($S_0 \rightarrow S_1$, $S_1 \rightarrow S_2$, $S_2 \rightarrow S_3$, and $S_{3,4} \rightarrow S_0$) Characterized by X-ray Absorption Spectroscopy at 20 K and Room Temperature† . *Biochemistry* **2005**, *44* (6), 1894-1908.